

Minor Isoflavones from the Roots of *Pueraria mirifica*

John L. Ingham*, Satoshi Tahara**, and Stanley Z. Dziedzic*

* Department of Food Science, Food Studies Building, University of Reading, Whiteknights, P.O. Box 226, Reading RG6 2AP, England

** Department of Agricultural Chemistry, Faculty of Agriculture, Hokkaido University, Kita-ku, Sapporo 060, Japan

Z. Naturforsch. **44c**, 724–726 (1989); received May 1989

Leguminosae, Papilionoideae, *Pueraria*, Isoflavonoids, Isoflavones

The isoflavone aglycone kwakhurin hydrate, and the glycosides genistin (genistein-7-O-glucoside) and puerarin-6''-monoacetate have been isolated from a methanolic extract of *Pueraria mirifica* roots.

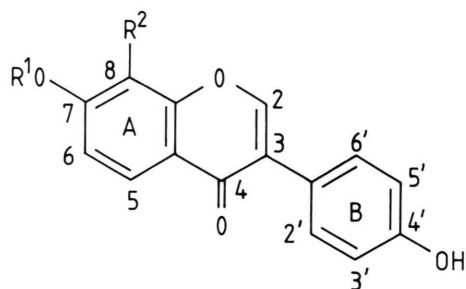
Introduction

In previous papers we described the isolation of three isoflavone aglycones (daidzein, genistein and kwakhurin) [1, 2], three isoflavone glycosides (daidzin **1**, puerarin **2**, and mirificin) [1, 3], and four coumestans (coumestrol, mirificoumestan, mirificoumestan hydrate and mirificoumestan glycol) [4] from roots of the Thai forest legume *Pueraria mirifica* Airy Shaw & Suvatabandhu. In this final paper of the series we report the isolation of three further isoflavones, genistin **3** (genistein-7-O-glucoside), puerarin-6''-monoacetate **4**, and kwakhurin hydrate **5**, from *P. mirifica* roots. Isoflavones **4** and **5** are recognized for the first time as natural products.

Results and Discussion

A methanolic extract of dried *P. mirifica* root was fractionated by Si gel TLC (Merck, F-254, layer thickness 0.5 mm) in CHCl₃–MeOH (CM, 20:1) as described earlier [1] to afford numerous fluorescence-quenching bands, many of which exhibited a blue or light blue fluorescence under long wavelength (365 nm) UV light. Apart from various known isoflavonoid aglycones [1, 2, 4], a minor component at approx. *R_f* 0.08 was identified as kwakhurin hydrate (**5**) by UV, MS and Si gel TLC comparison with authentic material previously obtained by treatment of kwakhurin with HCOOH [2].

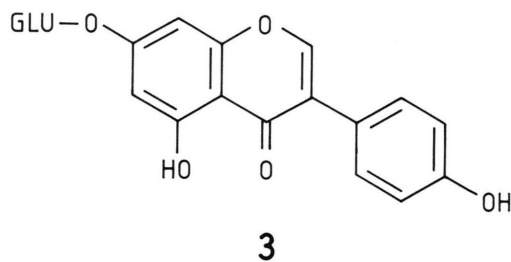
As expected, elution (MeOH) and further Si gel TLC (CHCl₃–MeOH–H₂O, CMH, 20:10:1) of the origin zone (*R_f* 0.00–0.05) from CM chromatograms yielded daidzin (**1**, daidzein-7-O-glucoside), puerarin



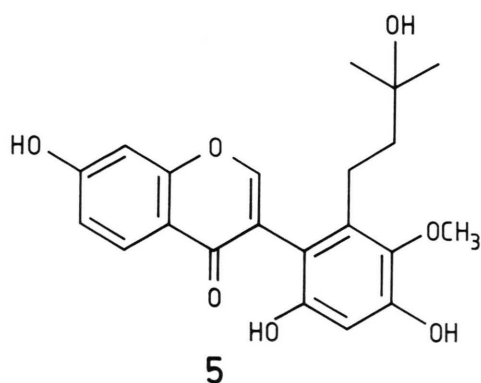
1: R¹ = GLU; R² = H

2: R¹ = H; R² = GLU

4: R¹ = H; R² = GLU-6''-OAc



3



5

Reprint requests to Dr. J. L. Ingham.

Verlag der Zeitschrift für Naturforschung, D-7400 Tübingen
0341–0382/89/0900–0724 \$ 01.30/0



Dieses Werk wurde im Jahr 2013 vom Verlag Zeitschrift für Naturforschung in Zusammenarbeit mit der Max-Planck-Gesellschaft zur Förderung der Wissenschaften e.V. digitalisiert und unter folgender Lizenz veröffentlicht: Creative Commons Namensnennung-Keine Bearbeitung 3.0 Deutschland Lizenz.

Zum 01.01.2015 ist eine Anpassung der Lizenzbedingungen (Entfall der Creative Commons Lizenzbedingung „Keine Bearbeitung“) beabsichtigt, um eine Nachnutzung auch im Rahmen zukünftiger wissenschaftlicher Nutzungsformen zu ermöglichen.

This work has been digitalized and published in 2013 by Verlag Zeitschrift für Naturforschung in cooperation with the Max Planck Society for the Advancement of Science under a Creative Commons Attribution-NoDerivs 3.0 Germany License.

On 01.01.2015 it is planned to change the License Conditions (the removal of the Creative Commons License condition “no derivative works”). This is to allow reuse in the area of future scientific usage.

(**2**, daidzein-8-C-glucoside) and mirificin (puerarin-6''-apioside) [1, 3]. However, upon TLC in benzene–MeOH (BM, 20:3, \times 3) the 'daidzin' material gradually separated into two compounds, pure daidzin **1** (major component, lower zone) and genistin **3** (genistein-7-O-glucoside; minor component, upper zone). The identification of genistin was based on a comparison (UV, TLC) with synthetic material. Genistin has previously been isolated from the flowers of *Pueraria thunbergiana* (= *P. lobata*) [5].

In addition to the four glycosides already mentioned, the CMH chromatogram revealed another compound running (*R_f* 0.69) above the daidzin/genistin band (*R_f* 0.63). This compound was immediately recognized as a 5-deoxyisoflavone from its typical long wavelength (365 nm) UV fluorescence on Si gel chromatograms (light blue, intensifying when fumed with NH₃ vapour [6]), and its UV spectrum in MeOH which resembled that of puerarin (**2**). A C-7 OH group was evident from the NaOAc-induced bathochromic shift of the methanolic UV maximum at 250 nm [6]. Treatment of the glycoside in MeOH with aqueous NH₃ caused the rapid formation of puerarin, a result consistent with acetylation of an aromatic (C-4') and/or sugar OH group. Monoacetylation was favoured from the detection of an intense FD-MS ion at *m/z* 458 (*M*⁺, base; *cf.* puerarin, *M*⁺ 416).

A ¹H NMR study indicated that ring B of puerarin and the new glycoside were identical (4'-OH) since virtually co-incident chemical shift values were in each case obtained for H-2'/6' and H-3'/5' (Table I). Similarly, the sugar H-1'' was evident as a doublet at δ 5.09 in the ¹H NMR spectrum of both compounds, whilst the 2''–5'' protons appeared as a complex series of signals between δ 3.40 and 4.19. In contrast, the 6'' protons (–CH₂OH) appeared at δ 3.75 dd and 3.88 dd in the spectrum of puerarin, shifting by 0.47 and 0.58 ppm to δ 4.22 dd and 4.46 dd respectively in that of the acetylated compound (OAc signal at δ 2.03 s). These shifts to significantly lower field confirm that the new glycoside is acetylated on the sugar unit at C-6''. In view of its similarity to puerarin, we suggest that the compound should be named puerarin-6''-monoacetate (**4**).

In a previous study, Bhutani *et al.* [7] isolated puerarin-4',6''-di-O-acetate from the roots of *Pueraria tuberosa*. However, no evidence was obtained to suggest that the latter compound was present in *P. mirifica* root extracts. Attempts to detect

Table I. Comparative ¹H NMR data (δ values) for puerarin (**2**) and puerarin-6''-monoacetate (**4**)^a.

Proton	Puerarin (2)	Puerarin-6''-monoacetate (4)
H-2	8.14 s	8.13 s
H-5	8.01 d (<i>J</i> = 8.8 Hz)	8.01 d (<i>J</i> = 8.8 Hz)
H-6	6.94 d (<i>J</i> = 8.8 Hz)	6.94 d (<i>J</i> = 8.8 Hz)
H-2'/H-6' [2H]	7.36 d (<i>J</i> = 8.8 Hz)	7.35 d (<i>J</i> = 8.8 Hz)
H-3'/H-5' [2H]	6.84 d (<i>J</i> = 8.8 Hz)	6.83 d (<i>J</i> = 8.8 Hz)
H-1''	5.09 d (<i>J</i> = 9.9 Hz)	5.09 d (<i>J</i> = 9.9 Hz)
H-2''	$\left\{ \begin{array}{l} 3.40\text{--}3.60\text{m [3H]}^b \\ 4.17\text{m}^b \end{array} \right.$	$\left\{ \begin{array}{l} 3.46\text{--}3.57\text{m [2H]}^c \\ 3.62\text{--}3.75\text{m}^c \\ 4.19\text{br t}^d \\ (J = ca. 10\text{ Hz}) \end{array} \right.$
H-3''		
H-4''		
H-5''		
H-6'' α , β	$\left\{ \begin{array}{l} 3.75\text{dd} \\ (J = 12.1, 4.8\text{ Hz}) \\ 3.88\text{dd} \\ (J = 12.1, 1.8\text{ Hz}) \end{array} \right.$	$\left\{ \begin{array}{l} 4.22\text{dd} \\ (J = 12.1, 5.9\text{ Hz}) \\ 4.46\text{dd} \\ (J = 12.1, 1.8\text{ Hz}) \end{array} \right.$
OAc [3H]	–	2.03 s

^a Spectra were determined in MeOH-*d*₄ at 270 MHz (TMS reference). Except where indicated by [2H] or [3H] the signals integrated for 1 proton. Coupling constants (*J*) are given in parentheses.

^b 3H and 1H signals probably H-3''/H-4''/H-5'', and H-2'' respectively.

^c 2H and 1H signals probably H-3''/H-4'', and H-5'' respectively. The shift of the H-5'' signal to lower field ($\Delta\delta$ ~0.2 ppm) in **4** when compared with **2** is consistent with the influence of a 6''-O-acetyl substituent, *cf.* ¹H NMR data in MeOH-*d*₄ for mosesin-2 with a β -galactoside residue (sugar H-5 at δ 3.49, and H-6 at δ 3.67 and 3.72), and its 6-acetyl derivative mosesin-1 (sugar H-5 at δ 3.71 [$\Delta\delta$ 0.22 ppm], and H-6 at δ 4.14 [$\Delta\delta$ 0.47 ppm] and 4.35 [$\Delta\delta$ 0.63 ppm]) [11].

^d 1H signal, probably H-2'', virtually unaffected by acetylation.

daidzein-7,4'-di-O-glucoside, earlier found in *P. thunbergiana* (*P. lobata*) [8], were similarly unsuccessful.

Experimental

Si gel TLC separations were carried out on Merck pre-coated plates (F-254, layer thickness 0.5 mm) using the following solvent systems: BM = benzene–MeOH, 5:2 or 20:3; CM = CHCl₃–MeOH, 20:1; CMW = CHCl₃–MeOH–H₂O, 20:10:1;

CMH = CHCl₃–MeOH–hexane (60–80 °C fraction from petroleum), 5:2:3; PEAM = *n*-pentane–diethyl ether–glacial acetic acid–MeOH, 75:25:6:20. MeOH was used as the eluting solvent.

Air-dried roots of *Pueraria mirifica* Airy Shaw & Suvatabandhu were extracted with aqueous MeOH as previously reported [1]. The extract was reduced to dryness *in vacuo* (35–40 °C) and the residue was then chromatographed (Si gel TLC) in CM to give various isoflavonoids which have been the subject of earlier papers [1, 2, 4]. In addition to these known compounds, the CM chromatogram also yielded kwakhurin hydrate (**5**, approx. *R_f* 0.08) which was eluted and further purified by Si gel TLC in PEAM (*R_f* 0.42). Pure **5** was finally obtained by TLC of the eluate from PEAM plates in BM (5:2, *R_f* 0.41). Elution of the near origin zone (*R_f* 0.00–0.05) from CM chromatograms gave a mixture of isoflavone glycosides which were separated by TLC in CMW [1]. In this solvent system puerarin-6''-monoacetate (**4**, *R_f* 0.69) was located immediately above the daidzin/genistin band (*R_f* 0.63) which in turn ran above puerarin (**2**, *R_f* 0.53). Si gel TLC of the daidzin + genistin eluate in BM (20:3, ×3) gave pure genistin (**3**, upper zone) just separated from daidzin (**1**, lower zone). Puerarin-6''-monoacetate from the CMW chromatogram was finally purified by TLC in CMH (*R_f* 0.35). In the same solvent system puerarin was located at *R_f* 0.15. In addition to their isolation from *P. mirifica* root material, further quantities of isoflavones **1**–**5** were also obtained by TLC of isoflavonoid-rich fractions remaining from an earlier investigation of the oestrogenic constituents of this plant [9, 10]. Kwakhurin hydrate (**5**) was isolated from these fractions by initial Si gel TLC in CM, 5:1 (*R_f* 0.46; *cf.* kwakhurin, *R_f* 0.62 [2]) followed by purification in PEAM and BM as described above.

Genistin **3** (genistein-7-*O*-glucoside)

UV data as lit. [6]. Hydrolysis in boiling MeOH with 2 *N* HCl [6] afforded genistein (5,7,4'-trihydroxyisoflavone) identical (UV, TLC) with an authentic sample.

Kwakhurin hydrate (**5**)

UV and MS data as lit. [2]. Long wavelength (365 nm) UV fluorescence on Si gel TLC plates, faint blue intensifying upon fuming with NH₃ vapour [6].

Puerarin-6''-monoacetate (**4**)

UV: λ_{max}, nm: MeOH 208, 243sh, 250, 262sh, 309sh; MeOH + NaOAc 260, 338 (addition of solid H₃BO₃ regenerated the MeOH spectrum). FD-MS: *m/z*, (%): 459 (M⁺ + 1; 50), 458 (M⁺; 100), 440 (28), 416 (M⁺ – 42; 15), 398 (21), 362 (20), 326 (29). Comparative FD-MS data for puerarin (**2**) were as follows: *m/z* 417 (M⁺ + 1; 39), 416 (M⁺; 100), 398 (15), 327 (12), 326 (28). ¹H NMR data, see Table I. *Conversion of 4 to 2*: Conc. aqueous NH₃ (approx. 0.5 ml) was added to a solution of **4** (5 mg) in MeOH. After standing at room temperature (approx. 18 °C) for 2 h, the solution was diluted with MeOH (20 ml) and reduced to dryness *in vacuo* (40 °C). Si gel TLC of the residue afforded a product indistinguishable (UV, TLC) from authentic puerarin (**2**).

Acknowledgements

We thank Mr. K. Watanabe for MS measurements, and Mrs. Y. Misu for determining the ¹H NMR spectra. Thanks are also due to Dr. G. S. Pope for his interest in the *Pueraria* project, and for providing dried root material and isoflavonoid-rich *P. mirifica* extracts.

- [1] J. L. Ingham, S. Tahara, and S. Z. Dziedzic, *Z. Naturforsch.* **41c**, 403 (1986).
- [2] S. Tahara, J. L. Ingham, and S. Z. Dziedzic, *Z. Naturforsch.* **42c**, 510 (1987).
- [3] J. L. Ingham, K. R. Markham, S. Z. Dziedzic, and G. S. Pope, *Phytochemistry* **25**, 1772 (1986).
- [4] J. L. Ingham, S. Tahara, and S. Z. Dziedzic, *Z. Naturforsch.* **43c**, 5 (1987).
- [5] T. Kurihara and M. Kikuchi, *J. Pharm. Soc. Japan* **96**, 1486 (1976).
- [6] T. J. Mabry, K. R. Markham, and M. B. Thomas, *The Systematic Identification of Flavonoids*, Springer, Berlin 1970.
- [7] S. P. Bhutani, S. S. Chibber, and T. R. Seshadri, *Indian J. Chem.* **7**, 210 (1969).
- [8] C. C. Fang, M. Lin, C. M. Sun, H. M. Liu, and H. Y. Lang, *Chung-Hua I Hsueh Tsa Chih* **54**, 271 (1974); *Chem. Abstr.* **82**, 47672 (1975).
- [9] H. E. H. Jones and G. S. Pope, *J. Endocrinol.* **22**, 303 (1961).
- [10] G. S. Pope, *Lab. Practice* **8**, 416 (1959).
- [11] K. Tachibana, K. Nakanishi, and S. H. Gruber, *Symp. Papers*, pp. 545–552, 27th Symp., Chemistry of Natural Products, Hiroshima 1985.